SCC-DFTB Study of Chrysotile Nanotube

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Lizardite and chrysotile are the most abundant minerals of the serpentine group and share the same chemical formula $Mg_3Si_2(OH)_2$. Lizardite has a flat-layered structure and chrysotile, the less abundant, occurs as cylindrically or spirally wrapped nanotubes (Fig.1)[1]. Recents studies indicate that chrysotile could be used as support for the immobilization of metalloporphyrins as well as adsorbent of ions and molecules[2]. Chrysotile has been the target for designing advanced materials with enhanced properties. In this work the self consistent charge – Density Functional Tight Binding (SCC-DFTB) method was used to study stability, electronic and mechanical properties of chrysotile. The Slater-Koster parameters for MgX (X = Mg, Si, O, H) have been calculated and the SCC-DFTB calculations compared with respect to the DFT calculations. The bond lengths and angles are in good agreement with DFT results, with an error around 0.05Å and 3 degrees, respectively. Chrysotile nanotubes with different quiralities and sizes have been calculated and their stability compared with respect to the lizardite. The SCC-DFTB calculations were performed using the deMon-nano and DFTB+ codes.

Keywords: Clay Mineral, DFT, SCC-DFTB, electronic and mechanical properties.

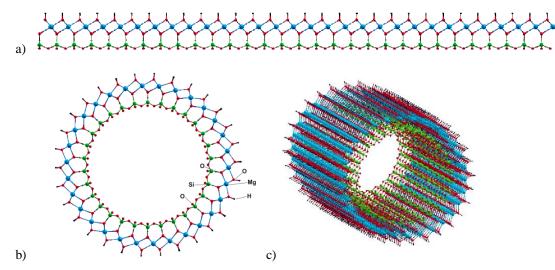


Figure 1. Structures of lizardite (2D) (a), chrysotile (2D) (b) and (3D) (c).

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